The Isolation of Terephthaldinitrile N-Oxide

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We wish to report the isolation and characterization of terephthaldinitrile N-oxide (I) as a stable compound. Although the chemistry of nitrile N-oxides has extensively been studied by several groups of workers in recent years, 1-3) few nitrile N-oxides4-6) have been isolated in a pure state because of the ease of their dimerization to furoxans.3,6,7)

Terephthaldihydroxamic chloride (II), the starting material in this work, has been used by several previous investigators, but no dinitrile N-oxide has been isolated.7-9)

To a solution of 0.233 g. of II⁸) (m. p. 185°C) in 20 ml. of methanol, an equivalent amount of triethylamine in 20 ml. of methanol was added at room temperature. I was soon precipitated out of the mixture as colorless needles. It was filtered and washed thoroughly with methanol (a 70% yield). When the sample was heated gradually, it showed a definite m. p. at 241~243°C (decomp.). On the other hand, when it was put in a bath heated at 150°C, it decomposed instantly.

Found: C, 59.79; H, 2.66; N, 17.77. Calcd.

¹⁾ a) R. Huisgen, Proc. Chem. Soc., 1961, 357; b) R. Huisgen, Angew. Chem., 75, 604 (1963); c) R. Huisgen, ibid., 75, 742 (1963).

²⁾ a) N. Barbulescu and P. Grünanger, Gazz. chim. ital., 92, 138 (1962); b) N. Barbulescu and A. Quilico, Gazz. chim. ital., 91, 326 (1961), and many previous papers.

³⁾ C. Grundmann, Angew. Chem., 75, 450 (1963).4) H. Wieland, Ber., 40, 1667 (1907).

⁵⁾ S. Califano, R. Scarpati and G. Speroni, Atti accad. nazl. Lincei Rend., Classe sci. fis. mat. e nat., 23, 263 (1958); Chem. Abstr., 52, 12557 (1958).

⁶⁾ S. Califano, R. Moccia, R. Scarpati and G. Speroni, J. Chem. Phys., 26, 1777 (1957).

⁷⁾ R. H. Wiley and B. J. Wakefield, J. Org. Chem., 25, 546 (1960).

⁸⁾ M. S. Chang and A. J. Matuszko. ibid., 28, 2260 (1963).

⁹⁾ A. Ricca, Gazz. chim. ital., 91, 83 (1961).

for $C_8H_4N_2O_2$: C, 60.00; H, 2.52; 2.52; N, 17.50.

Infrared spectra were obtained on Nujol mulls and a KBr disc: 841(m), 1025(w), 1100 (s), 1284(w), 1350(s), 1405(w) and 2330(s) cm⁻¹. Absorptions at 1350 and 2330 cm⁻¹ have been reported to be characteristic of arylnitrile N-oxides.7) No bands corresponding to furoxans10) were detected. I is soluble or slightly soluble in dioxane, acetonitrile, chloroform, ethylene dichloride, o-dichlorobenzene and m-cresol, but insoluble in benzene, diethyl ether, ligroin and carbon tetrachloride.

By the addition reaction of 50 mg. of I with methyl methacrylate at room temperature, 40 mg. of 1, 4-bis(5'-methyl-5'-methoxycarbonyl- $3'-\Delta^{2'}$ -isoxazolinyl) benzene¹¹⁾ was separated as a crystalline compound from the reaction mixture; m. p. 172~173°C (from methanol).

Found: C, 60.16; H, 5.48; N, 8.03. Calcd. for $C_{18}H_{20}N_2O_6$: C, 59.99; H, 5.59; N, 7.77. $\nu_{C=0}$, 1728(s) cm⁻¹ (Nujol).

By treating I with allyl alcohol in tetrahydrofuran, 1, 4-bis(5'-hydroxymethyl-3'-\(\Delta^2\)'-isoxazolinyl)benzene¹²⁾ was obtained; m. p. 214~ 216°C(decomp.) (from ethanol).

Found¹³⁾: C, 61.16; H, 5.98; N, 10.33. Calcd. for $C_{14}H_{16}N_2O_4$: C, 60.86; H, 5.84; N, 10.14.

Infrared absorptions: 3150~3400 cm⁻¹ region and 1046, 1055 cm⁻¹ doublet (primary alcohol).

Further studies are in progress on addition reactions of I with various mono- and difunctional unsaturated compounds.

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N. E. Boyer, G. M. Czerniak, H. S. Gutowsky and H. R. Snyder, J. Am. Chem. Soc., 77, 4238 (1955).
A. Quilico, G. S. D'Alcontres and P. Grünanger, Gazz. chim. ital., 80, 479 (1950); Chem. Abstr., 45, 3836 (1951). In this report, methyl 3-phenyl-5-methyl-2-isoxazoline-5carboxylate is described.

¹²⁾ G. S. D'Alcontres and P. Grünanger, Gazz. chim. ital., 80, 741 (1950); Chem. Abstr., 46, 971 (1952). Here 3phenyl-5-hydroxymethyl-2-isoxazoline is described.

¹³⁾ The authors are indebted for the microanalyses to the staff of the Analytical section of the Department of Organic Chemistry, Tokyo Institute of Technology.